# Structure of Puromycin Aminonucleoside 

By N. Padmaja, S. Ramakumar and M. A. Viswamitra<br>Department of Physics and ICMR Centre on Cell Biology, Indian Institute of Science, Bangalore-560 012, India

(Received 13 April 1988; accepted 1 July 1988)


#### Abstract

Amino- ${ }^{\prime}$ '-deoxy- $\mathrm{N}, \mathrm{N}$-dimethyladenosine, $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{O}_{3}, \quad M_{r}=294 \cdot 3$, monoclinic, $\quad P 2_{1}, \quad a=$ 4.684 (1), $b=10.252$ (1), $\quad c=14.381$ (3) $\AA, \quad \beta=$ 91.16 (2) ${ }^{\circ}, V=690.49 \AA^{3}, Z=2, D_{x}=1.39 \mathrm{Mg} \mathrm{m}^{-3}$, $\lambda(\mathrm{Cu} K \alpha)=1.5418 \AA, \mu=0.84 \mathrm{~mm}^{-1}, F(000)=312$, $T=295 \mathrm{~K}, R=0.045$ for 986 observed reflections with $I>1.5 \sigma(I)$. The adenine base is unprotonated at $\mathrm{N}(1)$ and is dimethylated at $\mathrm{N}(6)$. The nucleoside has a $3^{\prime}$-deoxy-3'-aminofuranosyl ribose sugar. The base is in the anti conformation $\left(\chi=2.3^{\circ}\right)$ with respect to the furanosyl ring, which shows a $\mathrm{C}\left(3^{\prime}\right)$-endo- $\mathrm{C}\left(2^{\prime}\right)$-exo pucker $\left({ }_{2} T^{3}\right)$. O( $5^{\prime}$ ) is disordered. The structure of the compound is similar to that of the nucleoside portion of the antibiotic puromycin.


Introduction. We report here the crystal structure of puromycin aminonucleoside, which forms the nucleoside portion of puromycin, a broad spectrum antibiotic and a structural analog of the $3^{\prime}$-aminoacyl adenosine moiety of charged tRNA. The present study was undertaken because of our interest in the conformational flexibility of modified nucleosides (Viswamitra \& Gautham, 1984). Our analysis shows that the conformational features of the molecule are similar to those of the nucleoside moiety of puromycin dihydrochloride pentahydrate, the structure of which has previously been solved (Sundaralingam \& Arora, 1972).

Experimental. Compound [6-dimethylamino-9-[(3'-amino- $3^{\prime}$-deoxyribosyl)purine] purchased from Sigma Chemicals. Crystals were obtained by direct evaporation from aqueous solution of the compound. Cell parameters were refined by least squares using 20 reflections, $3 \leq \theta \leq 28^{\circ}$, on a CAD-4 Enraf-Nonius diffractometer. Intensity data were collected up to $(\sin \theta) / \lambda=0.56 \AA^{-1}$, with $\omega-2 \theta$ scan using Ni-filtered $\mathrm{Cu} K \alpha$ radiation on a crystal of dimensions $0.2 \times$ $0.04 \times 0.03 \mathrm{~mm}$. Three standard reflections monitored at regular intervals, crystal stable to X-rays. Analytical absorption correction (North, Phillips \& Mathews, 1968) applied, transmission factor varied from 75.7 to $95.7 \%$. Index range $-5 \leq h \leq 5,0 \leq k \leq 11,0 \leq$ $l \leq 16$; of the 1181 unique reflections, 968 were considered observed $[I>1 \cdot 5 \sigma(I)$ ]. Structure solution by direct methods using MULTAN80 (Germain, Main \& Woolfson, 1971). The terminal oxygen atom, O(5'),
was found to be disordered. Of the 18 H atoms, 16 could be located from a difference electron density map. Number of parameters refined is 262. Full-matrix least-squares refinement on $F$ 's using SHELX76 (Sheldrick, 1976), with non-H atoms refined anisotropically and H atoms refined isotropically, converged at $R=0.045$. Individual weights $w \propto 1 /\left[\sigma^{2}(F)+\right.$ $0.001\left(F^{2}\right)$ ], $w R=0.042$; maximum $\Delta / \sigma=0.12 ; \Delta \rho$ in the final difference map within +0.22 and $-0.20 \mathrm{e}^{-3}$. The atomic scattering factors are as supplied in SHELX76. Structure solution using the Enraf-Nonius (1979) SDP on a PDP 11/44 computer.

Discussion. The final positional parameters along with equivalent isotropic temperature factors for the non-H atoms are given in Table 1.* Bond lengths and bond angles are listed in Table 2. Fig. 1 shows the conformation of the puromycin aminonucleoside molecule along with the atomic numbering scheme. The crystal packing viewed down the $a$ axis is shown in Fig. 2.

The nucleoside has an adenine base, which is unprotonated at $N(1)$ and dimethylated at $N(6)$, and a $3^{\prime}$-amino- ${ }^{\prime}$ '-deoxyfuranosyl ribose sugar. The bond lengths and bond angles are in good agreement with the average values found in crystal structures containing unprotonated adenine bases (Taylor \& Kennard, 1982). The dimethylation of the base at $\mathrm{N}(6)$ is found to have little or no effect on the bond distances as was also seen in the structure of puromycin (Sundaralingam \& Arora, 1972).

The nucleoside shows anti conformation with $\chi$ $\left[\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)\right]=2.3(8)^{\circ}$ compared with $18^{\circ}$ found in the puromycin structure. An earlier proton magnetic resonance (PMR) study of the compound has shown that it probably has a $50: 50$ population of $s y n$ and anti conformers in solution (Narula \& Dhingra, 1984).

The least-squares plane for the adenine base shows that it is essentially planar. The substituent atoms $\mathrm{N}(6)$

[^0]© 1988 International Union of Crystallography

Table 1. Final fractional coordinates and temperature factors of the non-hydrogen atoms, with e.s.d.'s in parentheses

| $U_{\text {eq }}=\frac{1}{3} \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eq }}\left(\AA^{2}\right)$ |
| $\mathrm{N}(1)$ | -0.5644 (9) | 0.4268 (6) | 0.7686 (3) | 0.0408 (14) |
| C(2) | -0.4084 (13) | 0.4444 (7) | 0.6927 (4) | 0.0459 (20) |
| $\mathrm{N}(3)$ | -0.2398 (9) | 0.5418 (6) | 0.6706 (3) | 0.0396 (14) |
| C (4) | -0.2333 (10) | 0.6313 (6) | 0.7389 (3) | 0.0341 (15) |
| C(5) | -0.3804 (11) | 0.6286 (6) | 0.8213 (3) | 0.0363 (15) |
| C (6) | -0.5573 (11) | 0.5179 (6) | 0.8358 (3) | 0.0365 (16) |
| $\mathrm{N}(7)$ | -0.3174 (11) | 0.7390 (6) | 0.8732 (3) | 0.0509 (16) |
| $\mathrm{C}(8)$ | -0.1364 (14) | 0.8040 (7) | 0.8228 (4) | 0.0510 (20) |
| $\mathrm{N}(9)$ | -0.0802 (9) | 0.7450 (6) | 0.7400 (3) | 0.0378 (13) |
| N (6) | -0.7187 (10) | 0.4981 (6) | 0.9099 (3) | 0.0480 (16) |
| C(9) | -0.7049 (23) | 0.5798 (9) | 0.9925 (5) | 0.0725 (29) |
| $\mathrm{C}(10)$ | -0.8993 (16) | 0.3830 (8) | 0.9163 (5) | 0.0583 (24) |
| $\mathrm{O}\left(4^{\prime}\right)$ | $0 \cdot 2497$ (7) | 0.9042 (5) | 0.6971 (2) | 0.0423 (11) |
| $\mathrm{C}\left(\mathrm{I}^{\prime}\right)$ | 0.1123 (13) | 0.7884 (6) | 0.6673 (4) | 0.0392 (17) |
| $\mathrm{C}\left(2^{\prime}\right)$ | -0.0405 (12) | 0.8181 (6) | 0.5757 (3) | 0.0376 (16) |
| $\mathrm{O}\left(2^{\prime}\right)$ | 0.1526 (9) | 0.8105 (0) | 0.5029 (3) | 0.0507 (14) |
| C( ${ }^{\prime}$ ) | -0.1178(11) | 0.9623 (6) | 0.5911 (4) | 0.0343 (15) |
| $\mathrm{N}\left(3^{\prime}\right)$ | -0.1872 (12) | 1.0362 (6) | 0.5077 (3) | 0.0407 (15) |
| C(4) | 0.1384 (11) | 1.0139 (6) | 0.6440 (4) | 0.0383 (16) |
| C(5') | 0.0872 (14) | 1.1252 (7) | 0.7093 (4) | 0.0327 (20) |
| $\mathrm{O}\left(5^{\prime}\right)^{*}$ | 0.3434 (14) | 1.1608 (7) | 0.7536 (5) | 0.0664 (24) |
| O(5)* | -0.0997 (28) | 1.0999 (13) | 0.7789 (7) | 0.0547 (40) |

Table 2. Bond lengths ( $\AA$ ) and bond angles ( ${ }^{\circ}$ ) involving the non -H atoms

| $\mathrm{N}(1)-\mathrm{C}(2)$ | $1.338(7)$ | $\mathrm{N}(6)-\mathrm{C}(10)$ | $1.456(10)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{C}(2)-\mathrm{N}(3)$ | $1.316(9)$ | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ | $1.413(8)$ |
| $\mathrm{N}(3)-\mathrm{C}(4)$ | $1.344(7)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $1.517(8)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.383(6)$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | $1.539(9)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.423(8)$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $1.504(8)$ |
| $\mathrm{C}(6)-\mathrm{N}(1)$ | $1.344(8)$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $1.500(9)$ |
| $\mathrm{C}(5)-\mathrm{N}(7)$ | $1.384(8)$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | $1.450(7)$ |
| $\mathrm{N}(7)-\mathrm{C}(8)$ | $1.309(8)$ | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | $1.396(9)$ |
| $\mathrm{C}(8)-\mathrm{N}(9)$ | $1.366(8)$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | $1.400(7)$ |
| $\mathrm{N}(9)-\mathrm{C}(4)$ | $1.369(8)$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right)$ | $1.450(8)$ |
| $\mathrm{C}(6)-\mathrm{N}(6)$ | $1.335(7)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(9)$ | $1.464(7)$ |
| $\mathrm{N}(6)-\mathrm{C}(9)$ | $1.454(9)$ | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}(5)$ | $1.368(13)$ |
| $\mathrm{C}(6)-\mathrm{N}(1)-\mathrm{C}(2)$ | $119.1(5)$ | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $107.4(5)$ |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{N}(3)$ | $129.8(6)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | $100.1(4)$ |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(4)$ | $110.3(5)$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $102.9(4)$ |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $127.5(5)$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | $105.6(4)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $116.3(5)$ | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(9)$ | $108.8(5)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)$ | $109.9(5)$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(9)$ | $113.2(5)$ |
| $\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)$ | $104.5(5)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | $109.7(4)$ |
| $\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)$ | $113.3(6)$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | $108.6(4)$ |
| $\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}(4)$ | $106.1(5)$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right)$ | $115.6(5)$ |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{N}(6)$ | $117.5(5)$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right)$ | $113.4(5)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{N}(6)$ | $125.4(5)$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $116.6(5)$ |
| $\mathrm{C}(9)-\mathrm{N}(6)-\mathrm{C}(10)$ | $115.6(6)$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | $109.7(6)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{N}(1)$ | $117.1(5)$ | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $108.7(5)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(9)$ | $106.1(5)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(9)-\mathrm{C}(4)$ | $125.5(5)$ |
| $\mathrm{C}(6)-\mathrm{N}(6)-\mathrm{C}(9)$ | $123.2(6)$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(9)-\mathrm{C}(8)$ | $128.4(5)$ |
| $\mathrm{C}(6)-\mathrm{N}(6)-\mathrm{C}(10)$ | $120.9(5)$ | $\mathrm{O}\left(5^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}(5)$ | $105.8(7)$ |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{N}(7)$ | $133.7(5)$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}(5)$ | $115.3(7)$ |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{N}(9)$ | $126.4(5)$ |  |  |
|  |  |  |  |

and $\mathrm{C}\left(1^{\prime}\right)$ are displaced on opposite sides of the plane by +0.028 (5) and -0.023 (6) $\AA$ respectively, unlike the puromycin structure, where they are on the same side of the base plane, displaced by 0.198 and $0.056 \AA$ respectively.

The ribose is in $\mathrm{C}\left(3^{\prime}\right)$-endo- $\mathrm{C}\left(2^{\prime}\right)$-exo conformation $\left({ }_{2} T^{3}\right)$, with the atoms $\mathrm{C}\left(3^{\prime}\right)$ and $\mathrm{C}\left(2^{\prime}\right)$ displaced by -0.261 (5) and $0.359(5) \AA$ respectively from the three-atom plane defined by $\mathrm{C}\left(1^{\prime}\right), \mathrm{C}\left(4^{\prime}\right)$ and $\mathrm{O}\left(4^{\prime}\right)$. The ribose exhibits a similar pucker in puromycin. The pseudorotation parameters of the furanose ring in the present structure are $P=-2.9$ and $38.1^{\circ}$ (Altona \& Sundaralingam, 1972).

The $\mathrm{O}\left(5^{\prime}\right)$ atom is disordered between gauche-trans ( $\mathrm{O} 5^{\prime}$ ) and gauche-gauche (O5) conformations about the bond $C\left(4^{\prime}\right)-C\left(5^{\prime}\right)$ with $2: 1$ occupancy ratio for the two sites. The PMR study (Narula \& Dhingra, 1984) estimated $80 \%$ of the nucleoside molecules to be in gauche-gauche conformation about the $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ bond, a conformation also seen in puromycin.

The present study thus shows that the conformation of molecules is similar to that of the nucleoside portion in puromycin. The conformational flexibility of puromycin seems to lie mainly in its amino-acid segment, as suggested by theoretical studies (Yathindra \& Sundaralingam, 1973) and the PMR solution study on puromycin and its related compounds (Narula \& Dhingra, 1984).


Fig. 1. Atomic numbering scheme for the puromycin aminonucleoside molecule.


Fig. 2. The crystal packing viewed down the $a$ axis.


Fig. 3. A view perpendicular to the adenine base plane, indicating stacking interactions.

## Molecular packing

The crystal structure is stabilized by the intermolecular . hydrogen bond $\mathrm{O}\left(2^{\prime}\right)-\mathrm{H} \cdots \mathrm{N}\left(3^{\prime}\right)$ [2.821 (6) $\AA$ and $172 \cdot 4^{\circ}$ ]. There are no stacking interactions due to ring overlap of bases. However, the exocyclic atoms $\mathrm{N}(6)$ and $\mathrm{C}(10)$ of the base stack on top ( $\simeq 3.54 \AA$ ) of the adenine base of the molecule related by the $a$-cell translation, as shown in Fig. 3.

We thank the Department of Science and Technology and Biotechnology, Government of India, for financial support.

## References

Altona, C. \& Sundaralingam, M. (1972). J. Am. Chem. Soc. 94, 8205-8212.
Enraf-Nonius (1979). Structure Determination Package. EnrafNonius, Deift, The Netherlands.
Germain, G., Main, P. \& Woolfson, M. M. (1971). Acta Cryst. A27, 368-376.
narula, S. S. \& Dhingra, M. M. (1984). J. Biomol. Struct. Dyn. 2, 175-190.
North, A. C. T., Phillips, D. C. \& Mathews, F. D. (1968). Acta Cryst. A24, 351-359.
Sheldrick, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England.
Sundaralingam, M. \& Arora, S. K. (1972). J. Mol. Biol. 71, 49-70.
Taylor, R. \& Kennard, O. (1982). J. Am. Chem. Soc. 104, 3209-3212.
Viswamitra, M. A. \& Gautham, N. (1984). Proc. Indian Acad. Sci. 93, 261-269.
Yathindra, N. \& Sundaralingam, M. (1973). Biochim. Biophys. Acta, 308, 17-27.

# Dimerization Products of Substituted 1,2-Benzoquinones 

By William H. Watson* and Ante Nagl $\dagger$<br>Department of Chemistry, Texas Christian University, Fort Worth, TX 76129, USA

and Wolfgang Steglich and Bernd Ebert
Institut für Organische Chemie und Biochemie der Universität, Gerhard-Domagk-Strasse 1, D-5300 Bonn 1, Federal Republic of Germany
(Received 30 December 1987; accepted 18 July 1988)


#### Abstract

Tetrahydro-2,8-dimethyl-1,4-ethanonaphthalene-5,6,9,10-tetrone (1), $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{4}, M_{r}$ $=244.25$, orthorhombic, $F 2 d d, a=7.505$ (1), $b=$ 14.180 (2),$\quad c=44.548$ (5) $\AA, \quad V=4741$ (1) $\AA^{3}, \quad Z=$ $16, D_{x}=1.369 \mathrm{~g} \mathrm{~cm}^{-3}, \lambda(\mathrm{Mo} K \alpha)=0.71073 \AA, \mu=$ $0.94 \mathrm{~cm}^{-1}, \quad F(000)=2048, \quad T=293 \mathrm{~K}, \quad R=0.0496$, 933 reflections. 1,4,4a,8a-Tetrahydro-9,9-dihydroxy-2,8 -dimethyl-1,4-ethanonaphthalene-5,6,10-trione (2), $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}_{5}, M_{r}=262 \cdot 27$, monoclinic, $C c, a=9.174$ (1), $b=10.799$ (2), $c=12.557$ (2) $\AA, \beta=103.40(2)^{\circ}, V$ $=1210.2(5) \AA^{3}, Z=4, D_{x}=1.439 \mathrm{~g} \mathrm{~cm}^{-3}, \lambda(\mathrm{Cu} K \alpha)$ $=1.54178 \AA, \quad \mu=8.76 \mathrm{~cm}^{-1}, \quad F(000)=552, \quad T=$


[^1]0108-2701/88/122178-05\$03.00
$295 \mathrm{~K}, R=0.0320$, 819 reflections. 1,4,4a,8a-Tetra-hydro-2,8-di-tert-butyl-1,4-ethanonaphthalene-5,6,9,10 -tetrone (3), $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{4}, M_{r}=328.41$, triclinic, $P \overline{1}$, $a=8.786$ (1),$\quad b=9.112$ (1),$\quad c=12.387$ (1) $\AA, \quad \alpha=$ 101.24 (1) , $\quad \beta=92.40(1), \quad \gamma=114.73$ (1) ${ }^{\circ}, \quad V=$ 875.0 (2) $\AA^{3}, Z=2, D_{x}=1.247 \mathrm{~g} \mathrm{~cm}^{-3}, \lambda(\mathrm{Mo} K \alpha)=$ $0.71073 \AA, \mu=0.80 \mathrm{~cm}^{-1}, F(000)=352, T=292 \mathrm{~K}$, $R=0.0499,1598$ reflections. The bridged portions of the three molecules exhibit similar conformations although the keto function in (2) is replaced by a dihydroxy substituent. The cyclohexenedione ring in (1) is more flattened than that in (2) and (3). Bond $\mathrm{C}(1)-\mathrm{C}(8 a)=1.572(5)$ is significantly longer than the $\mathrm{C}(4)-\mathrm{C}(4 a)$ distance of 1.554 (1). It is suggested that this difference is inherent in the ring system and does not arise from steric factors.


[^0]:    * Lists of structure factors, anisotropic thermal parameters, torsion angles and H -atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51201 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH 1 2HU, England.

[^1]:    * Author to whom correspondence should be addressed.
    $\dagger$ On leave from Faculty of Technology, University of Zagreb, Zagreb, Yugoslavia.

